

Balancing Chemical Equations Teaching Transparency Worksheet

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Balanced chemical equations A balanced equation models a chemical reaction using the formulae of the reactants and products . It shows the number of units of each substance involved.

Balanced chemical equations - Introducing chemical ...

Instructions. To balance a chemical equation, enter an equation of a chemical reaction and press the Balance button. The balanced equation will appear above. Use uppercase for the first character in the element and lowercase for the second character. Examples: Fe, Au, Co, Br, C, O, N, F. Ionic charges are not yet supported and will be ignored.

Chemical Equation Balancer

Balance the following chemical equations. 1. $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2$: 2. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{CO}_2$: 3. $\text{KOH} + \text{H}_3\text{PO}_4 \rightarrow \text{K}_3\text{PO}_4 + \text{H}_2\text{O}$: 4. $\text{SnO}_2 + \text{H}_2 \rightarrow \text{Sn} + \text{H}_2\text{O}$: 5. $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO} + \text{H}_2\text{O}$: 6. $\text{KNO}_3 + \text{H}_2\text{CO}_3 \rightarrow \text{K}_2\text{CO}_3 + \text{HNO}_3$: 7. $\text{B}_2\text{Br}_6 + \text{HNO}_3 \rightarrow \text{B}(\text{NO}_3)_3 + \text{HBr}$: 8. $3 + \text{BF}_3 + \text{Li}_2\text{SO}_3 \rightarrow \text{B}_2(\text{SO}_3)_3 + \text{LiF}$: 9. $(\text{NH}_4)_3\text{PO}_4 + \text{Pb}(\text{NO}_3)_4 \rightarrow \text{Pb} \dots$

Balancing Equations: Practice Problems

A chemistry tutorial designed to help learn the basic principles of balancing chemical equations, along with examples and methods of balancing different chem...

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